

PILLAI COLLEGE OF ENGINEERING, NEW PANVEL
(Autonomous) (Accredited 'A+' by NAAC)
END SEMESTER EXAMINATION
SECOND HALF 2021

BRANCH: FE (ECS/EXTC)

Subject: Engineering Chemistry – I Solution
Max. Marks: 45

Time: 02.00 Hours
Date: 08-04-2022

1. A) Two functions- 2M
 One example-1M
- B) Moisture reduces 1) heating value
 2) Calorific Value3M
 3) Efficiency
- C) Indicator- EBT
 Buffer Solution- $\text{NH}_4\text{Cl} + \text{NH}_4\text{OH}$ 3M
 End point- Wine red to blue
- D) Viscosity (definition & Significance)
 Viscosity Index (definition & Significance).....3M
- E) Explanation of Passivity of Al.....3M

2. A) Four differences between Octane No. & Cetane No.- 2M
 Compound with highest Octane no. - 1M
 Compound with highest Cetane no. - 1M
- B) Description of each factor.....1x5 = 5M
- C)

Constituent	% by weight	Wt. of each per Kg of fuel
C	82	.82
H	3	.03
O	8	.08
S	2	.02
N	2	Doesn't participate
Ash	3	Doesn't participate

..1M

$$\begin{aligned} \text{Amount of air required (Kg)} &= 100/23(2.67C+ 8H+S-O) \\ &= 100/23(2.67 \times 0.82+ 8 \times 0.03+ 0.02-0.08) \\ &= 10.30 \text{ Kg} \dots\dots\dots 3M \end{aligned}$$

Amount of air required (Kg) for 2 Kg of Fuel= 20.60 kg

$$\begin{aligned} \text{Volume of air for 2kg} &= 20.60 \times 22.4 / 28.94 \\ &= 15.94 \text{ m}^3 \end{aligned}$$

3. A) Stress Corrosion

Definition.....1M
 Condition.....2M
 Example.....1M

B) Strength of SHW = 1mg/ml.....1M

Strength of EDTA= 50/48 mg of CaCO₃ eqv.....1M

Total hardness= 312.5 ppm.....1M

Permanent hardness= 208.33ppm.....1M

Temporary hardness= 104.17 ppm.....1M

C) Four differences with examples.....6M

4. A) four differences of scale & sludge

B) Definition- 1M

Reaction- 2M

Advantages- 1M

Disadvantages-1M

C) Cathodic Protection

Principle- 2M

Sacrificial Protection-2M

ICCP- 2M